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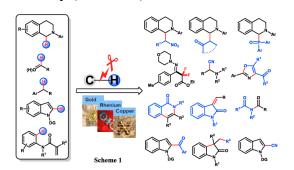
OR16 – Photoredox Catalyzed C–H Bond Functionalization and Access to Fluorinated Compounds

Pan Xu, Weipeng Li, Jin Xie and Chengjian Zhu*

State Key Laboratory of Coordination Chemistry, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210093, China

E-mail: cjzhu@nju.edu.cn

The C–H functionalization is a perennial topic of interest for organic chemists, it is a great challenging task to accomplish highly selective C–H bond functionalization under mild conditions. In the past a few years, the directed alkylation, cyanation, oxidation, amination, acylation and phosphonation of C–H bond were realized in our laboratory, with transition metal (Au, Re, Rh, Cu, etc.) catalysis or metal-free conditions. With those transformation methods, different kind of heterocyclic compounds were effectively constructed as shown in Scheme 1.¹ Owing to its significant advances in energy-saving and environmentally benign features, visible-light photoredox catalysis has witnessed rapid development and attracted considerable attention in both academia and industry. The incorporation of a fluorinated group into organic molecules is particularly intriguing because of its significant applications in the life sciences. Recently, our group achieved visible-light-induced trifluromethylation and difluroalkylation with different kind of fluro reagents, several kinds of fluorinated compounds were synthesized efficiently (Scheme 2).²



Visible-light Photoredox Catalysis Irippy)₃ or Ru(ppy)₃

References

- Chem. Soc. Rev., 2014, 43, 5245; Angew. Chem. Int. Ed. 2012,51, 1252; Angew. Chem. Int. Ed. 2013, 52,6288; Angew. Chem. Int. Ed. 2016, 55, ASAP (DOI: 10.1002/anie.201511917); Chem. Commun. 2016, 52, 447; Chem. Commun. 2015, 51, 1320; Chem. Commun. 2015, 51, 3786; Chem. Commun. 2013, 49, 819; Chem. Commun. 2013, 49, 2933; Chem. Commun, 2013,49, 3700; Chem. Eur. J. 2013, 19,1914; Chem. Eur. J. 2013, 19, 9754; Chem. Eur. J. 2013, 19, 11553; Chem. Eur. J. 2014, 20, 3606; ACS Catal. 2012, 2, 163; ACS Catal. 2013, 3, 1365; ACS Catal. 2013, 3, 2195; Org. Lett. 2013, 15, 4062; Org. Lett. 2016, 18,1143; Adv. Synth. Catal. 2015, 357, 1149; Adv. Synth. Catal. 2013, 355, 2029; Adv. Synth. Catal. 2012, 354, 1646.
- Angew. Chem. Int. Ed. 2016, 55, 2939; Chem. Sci. 2013, 4,1281; Chem. Commun. 2016, 52, 4761; Chem. Commun. 2015, 51, 13508; Chem. Commun., 2015, 51, 7222; Chem. Commun. 2014, 50, 2308; Chem. Commun. 2014, 50, 7521; Chem. Commun. 2013, 49, 5672; Org. Lett. 2014, 16, 1768; Chem. Eur. J. 2013, 19, 14039; Adv. Synth. Catal. 2015, 357, 1277; Adv. Synth. Catal. 2015, 357, 3057.